



Partnership for Clean Indoor Air

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Aprovecho Research Center
Advanced Studies in Appropriate Technology Laboratory

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USAID
FROM THE AMERICAN PEOPLE

SNV

Netherlands

Development

Connecting People's Capacities

Organisation

Stove Design and Performance Training

Vientianne Laos
March 14-16, 2011

Mike Hatfied - Sam Bentson - Ryan Thompson
Aprovecho Reserch Center, USA



Aprovecho Research Center

**Started in 1976 with invention
of Lorena Stove**

A failure!!!!



Aprovecho Stove Projects and Trainings



Mexico
Guatemala
Honduras
El Salvador
Nicaragua

In Field Projects

Ghana

Brazil
Bolivia

Uganda
Malawi
South Africa
Lesotho
Zaire

Indonesia

India
Philippines
Tibet
China

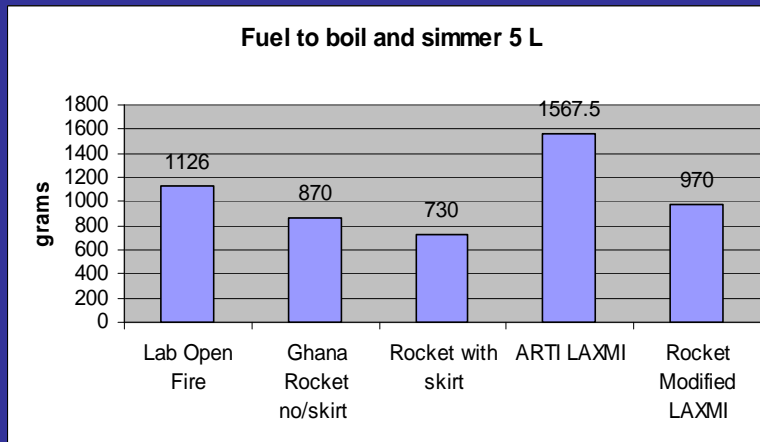
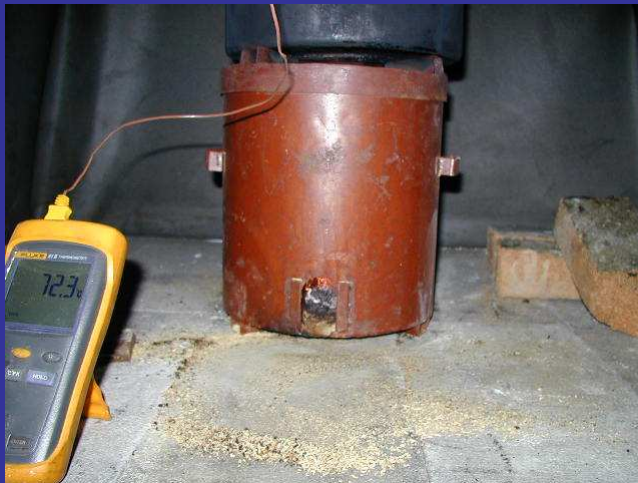
Active Stove Projects in over 20 countries – Trainings in 30 countries with participants from over 60 countries – Tested and evaluated over 100 stoves



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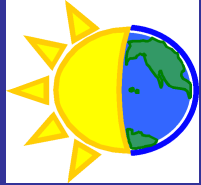
80574 Hazelton Road, Cottage Grove, OR 97424-9747, USA
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Lab Work



International Standards of Performance

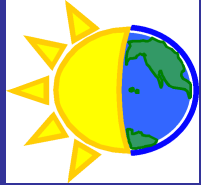
Creation of Regional Testing Centers



Elements of an improved stove (ICS)

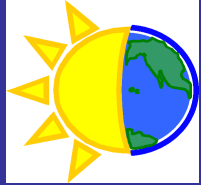
There are four goals we need to meet when designing a stove—

- 1- The stove cooks food as well or better than the traditional method
- 2- The stove eliminates or reduces the amount of smoke in the kitchen
- 3- It uses less fuel to cook food
- 4- Is producible at a cost that is acceptable to users



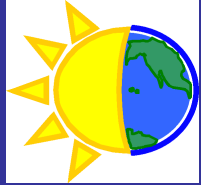
Elements of an improved stove

- 1- The stove cooks food as well or better than the traditional method
 - Use local cooks throughout design process
 - Form a stove committee
 - Perform tests using local cooks (CCT and KPT)
 - Follow up by independent organizations



Elements of an improved stove

- 2- The stove eliminates or reduces the amount of smoke in the kitchen
 - Worldwide 1.6 million people, mostly women and young children, die each year from breathing wood smoke!!
 - By cleaning up combustion as best we can and then making sure cooks are not exposed to what smoke is produced we reduce health risks of biomass cooking
 - Chimneys

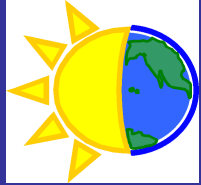


Elements of an improved stove

3- It uses less fuel to cook food

Often meeting goals one and two will be in direct conflict with fuel savings

Requires testing (WBT, CCT, KPT) to determine if fuel is in truth being saved



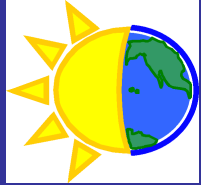
Elements of an improved stove

4- Is producible at a cost that is acceptable to users

Given enough money almost any stove can be made

To reach the majority of the some 2.6 billion people who cook on wood we need to have a stove that sells for less than \$5-\$10

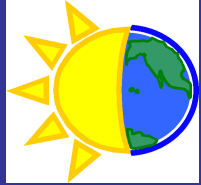
Other options – Micro Finance, Carbon Credits...?



Elements of an improved stove (ICS)

There are four goals we need to meet when designing a stove—

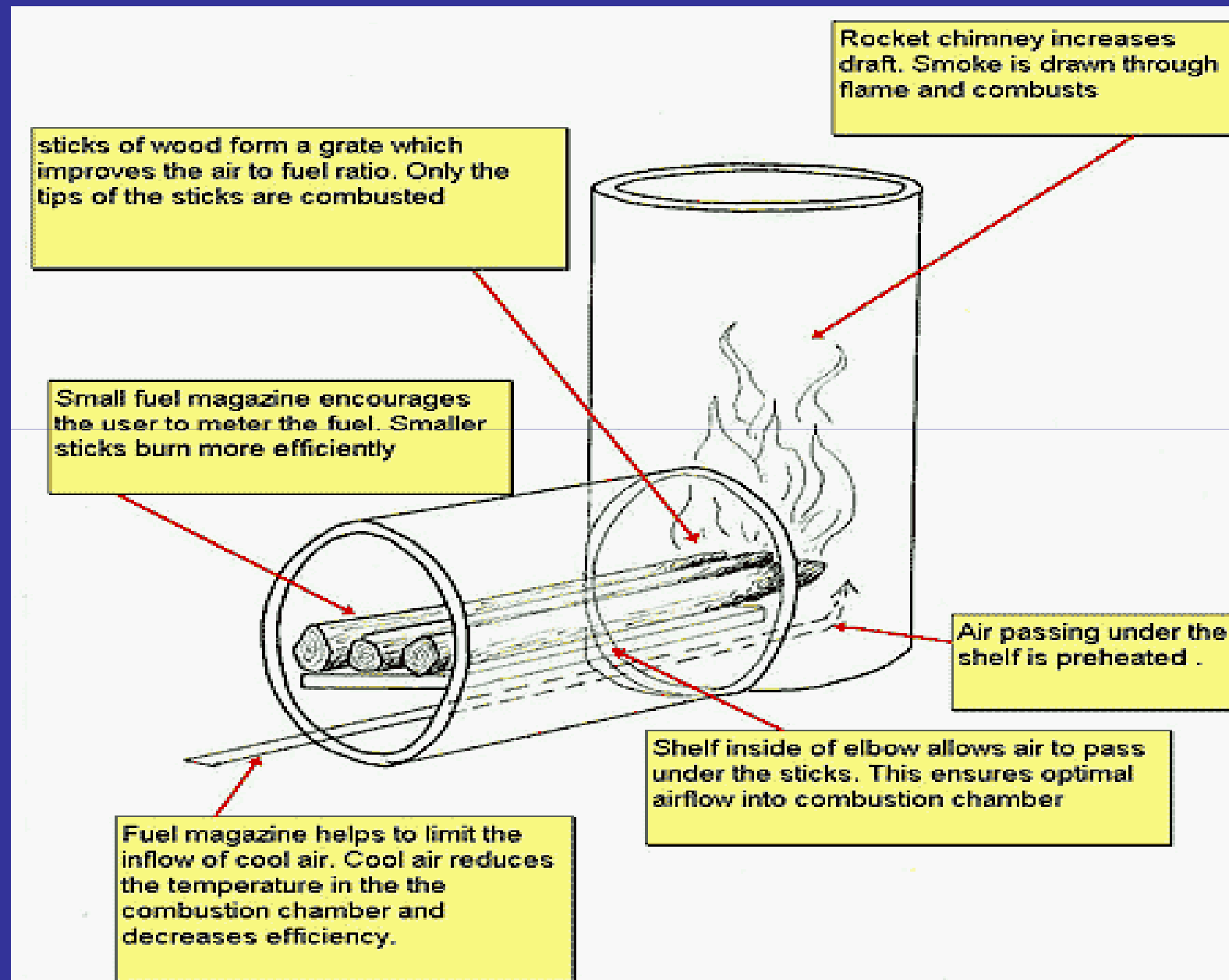
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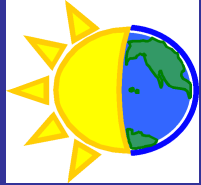


Simplified Designing of an Improved Stove

- 1 – Improved Combustion and/or chimney
- 2- Improved Heat Transfer

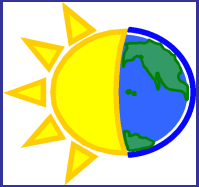
There are many ways to achieve our goals





Simplified combustion theory review

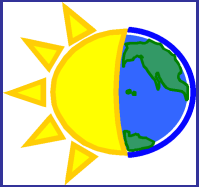
- Wood doesn't burn
- Wood gets hot and releases volatile gases that then combust
- If wood is heated to 650 degrees Celsius (and sufficient oxygen is mixed with the volatile gases) the result is complete combustion . The products of clean combustion are CO₂ , water vapour and heat.
- A lot of heat , roughly speaking , dry wood has **half** the energy per kg as gasoline,
- Smoke is wasted energy



What are limiting factors to high temperatures and achieving complete combustion ?

Challenge # 1

- **Cool stove body**
- **Cool earth**
- the body of the stove or of the earth **robs** heat from the fire
- which lowers combustion temperatures... which decreases efficiency...and increases smoke



What are limiting factors to high temperatures and achieving complete combustion ?

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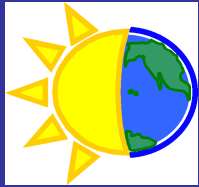
Solution?

- **Insulate the stove** with low mass, heat resistant materials in order to keep the fire as hot as possible
- Remember mass is the opposite of insulation
- Effective stove insulators are pumice , vermiculite, and wood ash
- Dense things such as earth,sand, cement, water and cast iron are poor insulators



Maximizing combustion efficiency

- Challenge #2
- Cool wood
 - which lowers combustion temperatures...which decreases efficiency...And increases smoke



Maximizing combustion efficiency

- Challenge #2

- Cool wood
- which lowers combustion temperatures...which decreases efficiency...And increases smoke

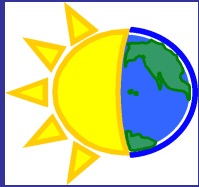
- Solution?

- Meter the fuel!
- Use small sticks whenever possible
- Maximize the surface area of the wood exposed to coals
- Heat only the fuel that is burning
- Burn the tips of sticks only as they enter the combustion chamber



Maximizing combustion efficiency

- Challenge # 3
- Cool air/ Too much air
 - which lowers combustion temperatures... which decreases efficiency... And increases smoke
- Note: an open fire can draw 20 times more than is required for stoichiometric (chemically ideal) combustion



Maximizing combustion efficiency

- Challenge # 3

- Cool air/ Too much air

- which lowers combustion temperatures... which decreases efficiency... And increases smoke

- Note: an open fire can draw 20 times more than is required for stoichiometric (chemically ideal) combustion

- Solution ?

- Do not allow too much or too little air to enter the combustion chamber.
- there should be a minimum excess of air supporting clean burning.



Maximizing combustion efficiency

- Challenge # 4
- Cool cooking pot
- The cooking pot is generally no more than a 100 –200 degrees Celsius
- Flames touching the pot?
- **Soot and smoke!**



Maximizing combustion efficiency

- Challenge # 4

- Cool cooking pot
- The cooking pot is generally no more than a 100 –200 degrees Celsius
- Flames touching the pot?
- **Soot and smoke!**

- Solution?

- Elevate the pot above the height of the flames
- This creates an internal ‘chimney’ which increases draft
- And gives time for improved air/ fuel mixing



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sticks of wood form a grate which improves the air to fuel ratio. Only the tips of the sticks are combusted

Small fuel magazine encourages the user to meter the fuel. Smaller sticks burn more efficiently

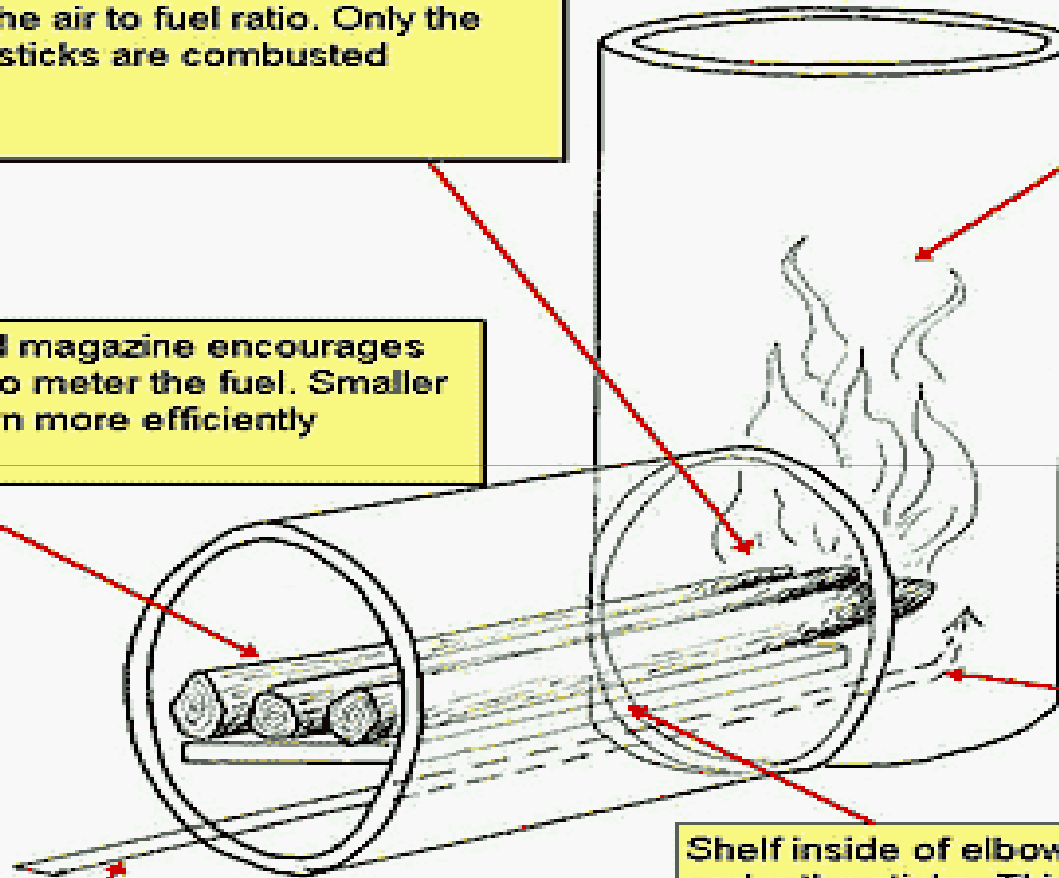
Fuel magazine helps to limit the inflow of cool air. Cool air reduces the temperature in the the combustion chamber and decreases efficiency.

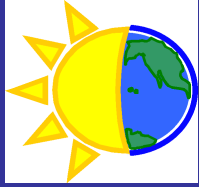
Rocket chimney increases draft. Smoke is drawn through flame and combusts

Stove
Wood
Air
Pot

Air passing under the shelf is preheated .

Shelf inside of elbow allows air to pass under the sticks. This ensures optimal airflow into combustion chamber



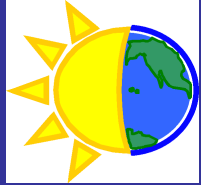


Complete combustion

Baldwin found that even a smoky fire can be as high as
92% Combustion Efficiency

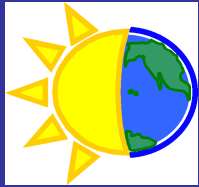
Worldwide 1.6 million people, mostly women and young
children, die each year from breathing wood smoke!!

If we did not care about health or user satisfaction we
would probably not work on CE but instead focus on
heat transfer to the pot



Simplified Designing of an Improved Stove

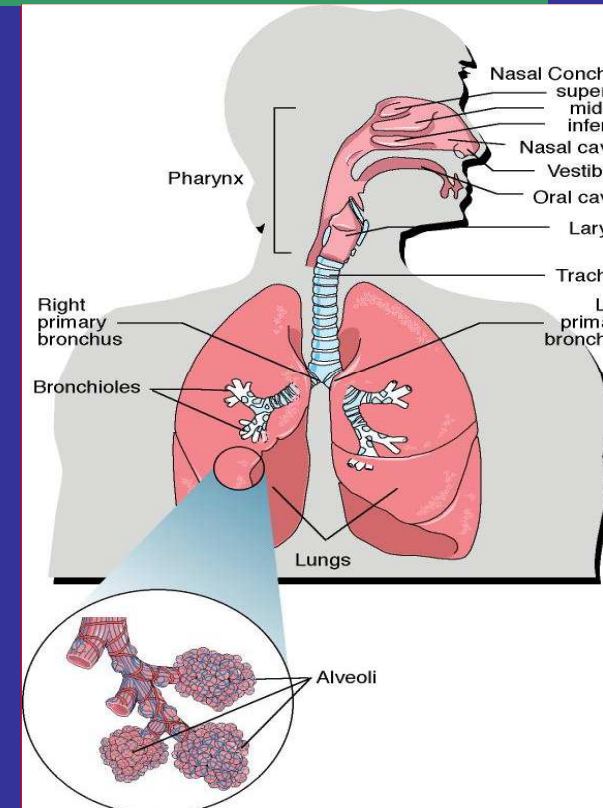
- 1 – Improved Combustion and/or chimney
 - Very little effect on Fuel use!!
 - Important for Human health



Complete combustion

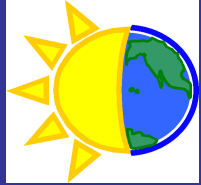
Carbon Monoxide (CO) -
Odorless and invisible

Particulate Matter (PM)
– Visible Smoke



A Few of the Chemicals in Woodsmoke (~g/kg emission factors)

• Carbon Monoxide	80-370	<i>Oxygenated PAHs</i>	0.15-1
• Methane	14-25	<i>Polycyclic Aromatic Hydrocarbons (PAH)</i>	
• VOCs (C2-C7)	7-27	Fluorene	4x10 ⁻⁵ - 1.7x10 ⁻²
• <i>Aldehydes</i>	0.65.4	Phenanthrene	2x10 ⁻⁵ - 3.4x10 ⁻²
– Formaldehyde	0.1-0.7	Anthracene	5x10 ⁻⁵ - 2.1x10 ⁻⁵
– Acrolein	0.02-0.1	Methylanthracenes	7x10 ⁻⁵ - 8x10 ⁻⁵
– Propionaldehyde	0.1-0.3	Fluoranthene	7x10 ⁻⁴ - 4.2x10 ⁻²
– Butryaldehyde	0.01-1.7	Pyrene	8x10 ⁻⁴ - 3.1x10 ⁻²
– Acetaldehyde	0.03-0.6	Benzo(a)anthracene	4x10 ⁻⁴ - 2x10 ⁻³
– Furfural	0.2-1.6 1.6	Chrysene	5x10 ⁻⁴ - 1x10 ⁻²
• <i>Substituted Furans</i>	0.15-1.7	Benzofluoranthenes	6x10 ⁻⁴ - 5x10 ⁻³
• Benzene	0.6-4.0	Benzo(e)pyrene	2x10 ⁻⁴ - 4x10 ⁻³
• <i>Alkyl Benzenes</i>	1-6	Benzo(a)pyrene	3x10 ⁻⁴ - 5x10 ⁻³
• Toluene	0.15-1.0	Perylene	5x10 ⁻⁵ - 3x10 ⁻³
• Acetic Acid	1.8-2.4	Ideno(1,2,3-cd)pyrene	2x10 ⁻⁴ - 1.3x10 ⁻²
• Formic Acid	0.06-0.08	Benz(ghi)perylene	3x10 ⁻⁵ - 1.1x10 ⁻²
• <i>Nitrogen Oxides</i> (NO,NO ₂)	0.2-0.9	Coronene	8x10 ⁻⁴ - 3x10 ⁻³
• Sulfur Dioxide	0.16-0.24	Dibenzo(a,h)pyrene	3x10 ⁻⁴ - 1x10 ⁻³
• Methyl chloride	0.01-0.04	Retene	7x10 ⁻³ - 3x10 ⁻²
• Napthalene	0.24-1.6	Dibenz(a,h)anthracene	2x10 ⁻⁵ - 2x10 ⁻³
• <i>Substituted Napthalenes</i>	0.3-2.1	<i>Trace Elements</i>	
• <i>Oxygenated Monoaromatics</i>	1 - 7	Cr	2x10 ⁻⁵ - 3x10 ⁻³
– Guaiacol (and derivatives)	0.4-1.6	Mn	7x10 ⁻⁵ - 4x10 ⁻³
– Phenol (and derivatives)	0.2-0.8	Fe	3x10 ⁻⁴ - 5x10 ⁻³
– Syringol (and derivatives)	0.7-2.7	Ni	1x10 ⁻⁶ - 1x10 ⁻³
– Catechol (and denvatives)	0.2-0.8	Cu	2x10 ⁻⁴ - 9x10 ⁻⁴
• Particulate Organic Carbon	2-20	Zn	7x10 ⁻⁴ - 8x10 ⁻³
• <i>Chlorinated dioxins</i>	1x10 ⁻⁵ - 4x10 ⁻⁵	Br	7x10 ⁻⁵ - 9x10 ⁻⁴
• Particulate Acidity	7x10 ⁻³ - 7x10 ⁻²	Pb	1x10 ⁻⁴ - 3x10 ⁻³
• <i>Normal alkanes</i> (C ₂₄ -C ₃₀)	1x10 ⁻³ - 6x10 ⁻³	Elemental Carbon	0.3 - 5
		<i>Cyclic di-and triterpenoids</i>	
		Dehydroabiatic acid	0.01 - 0.05
		Isopimaric acid	0.02 - 0.10
		Lupenone	2x10 ⁻³ - 8x10 ⁻³
		Friedelin	4x10 ⁻⁶ - 2x10 ⁻⁵

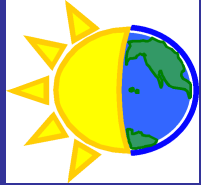


The second half of the equation

The true efficiency or fuel saving potential of a stove comes from two factors-

- Combustion Efficiency
- Heat Transfer Efficiency

$$\text{Total efficiency} = \text{CE} \times \text{HTE}$$



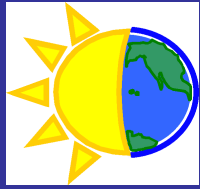
The second half of the equation

Total efficiency = CE x HTE

The simplest of rocket stoves, the insulated elbow alone, can be said to have an overall efficiency of about 18%

If we are getting above 90% CE, what is the HTE?

$$18\% = 90\% \times \text{HTE} \quad \text{--} \quad \text{HTE} = 20\%$$



The second half of the equation

$$\text{Total Eff} = 18\% = 90\% \times 20\%$$

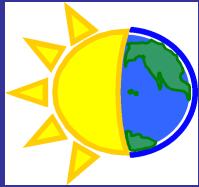
What happens if we work to get CE up to 100% (a difficult 10% increase)?

$$\text{Total E} = 100\% \times 20\% = 20\%$$

What happens if we work to raise HTE by 10% (a much easier increase)?

$$\text{Total E} = 90\% \times 30\% = 27\%$$

Always work on the weakest link!!!



Optimising heat transfer

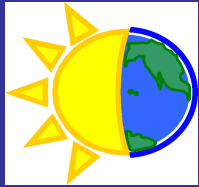
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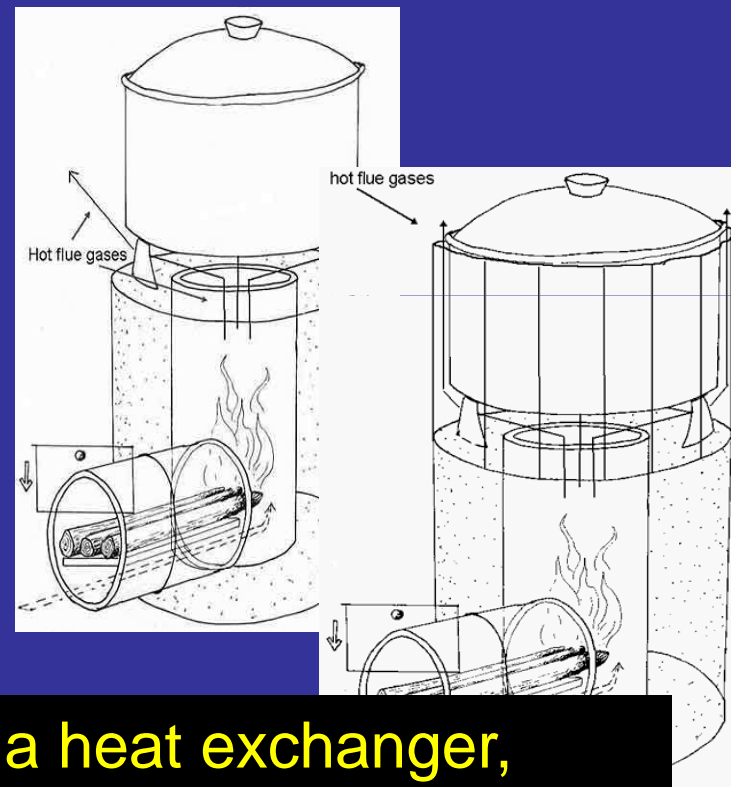
heat exchanger/skirt



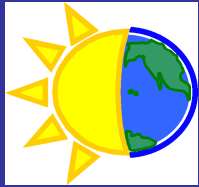


Optimising heat transfer

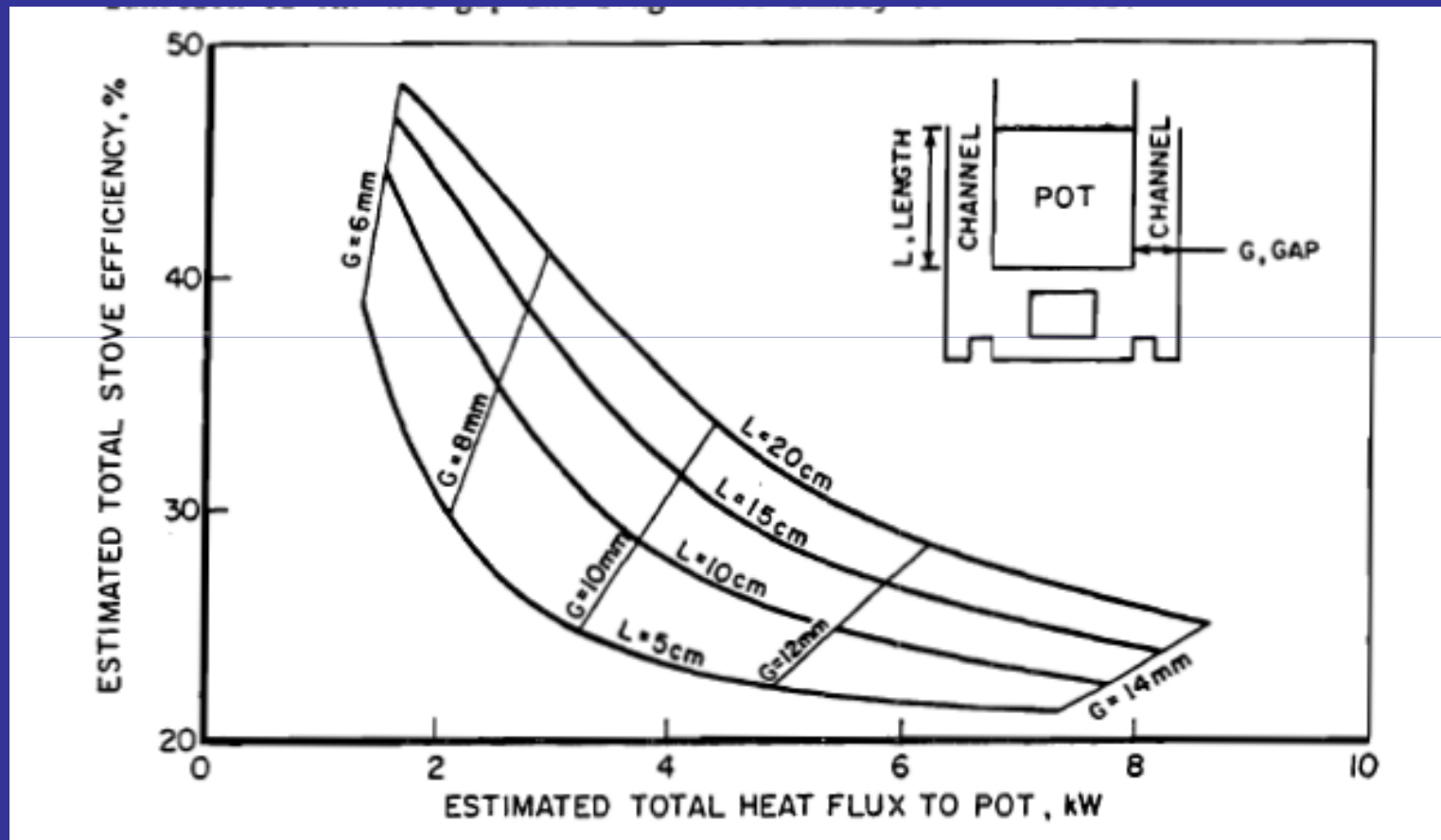
- Maximize **surface area** of pot that is exposed to hot flue gases
- Keep cross sectional area constant throughout flow path of hot gasses
- Maximize Temperature difference between hot gases and pot
- Maximize **velocity** of hot flue gases to disturb boundary layer



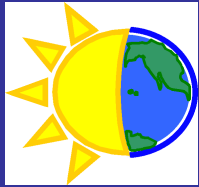
With a heat exchanger, overall efficiency can be improved by 50% or more



Optimising heat transfer



With a heat exchanger, overall efficiency can be improved by 50% or more



Optimising heat transfer

Complete Combustion

- Insulated combustion chamber
- Metered Fuel
- Metered/Preheated air
- Pot kept away from Combustion zone

Improved Heat Transfer

- Maximize **surface area**
- Constant cross sectional area
- Maximize temp difference between hot gases and pot (insulate against losses)
- Maximize **velocity** of hot flue gases

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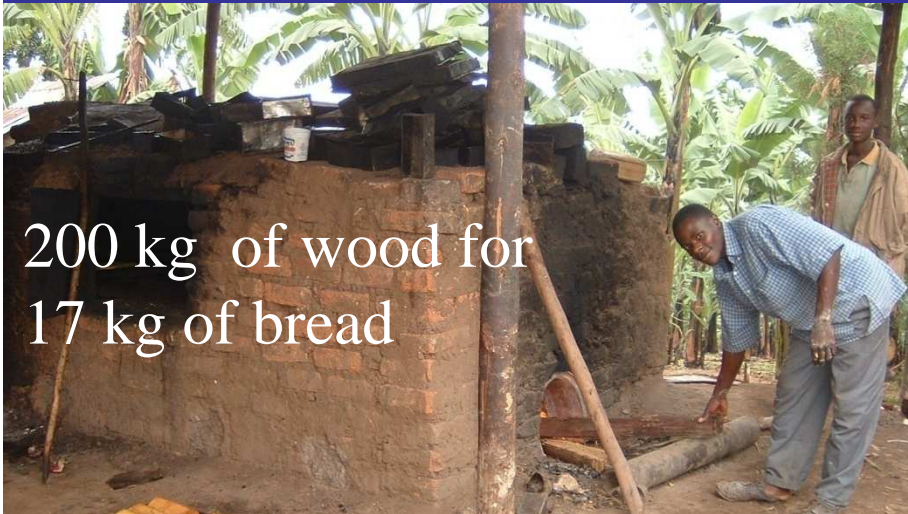
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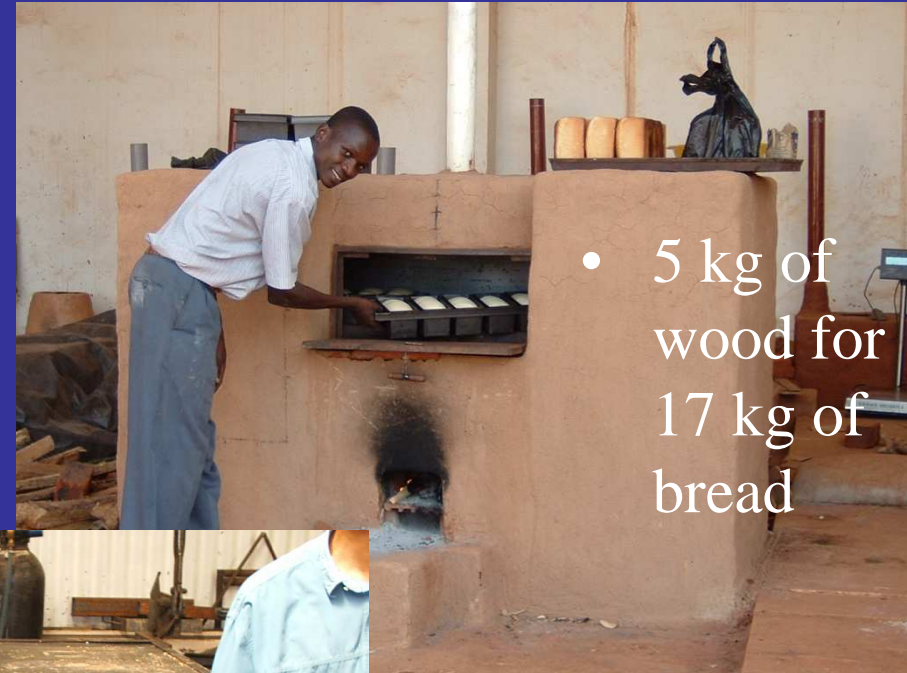
A few rocket stove design possibilities



Rocket Bread Oven



200 kg of wood for
17 kg of bread



- 5 kg of wood for
17 kg of bread



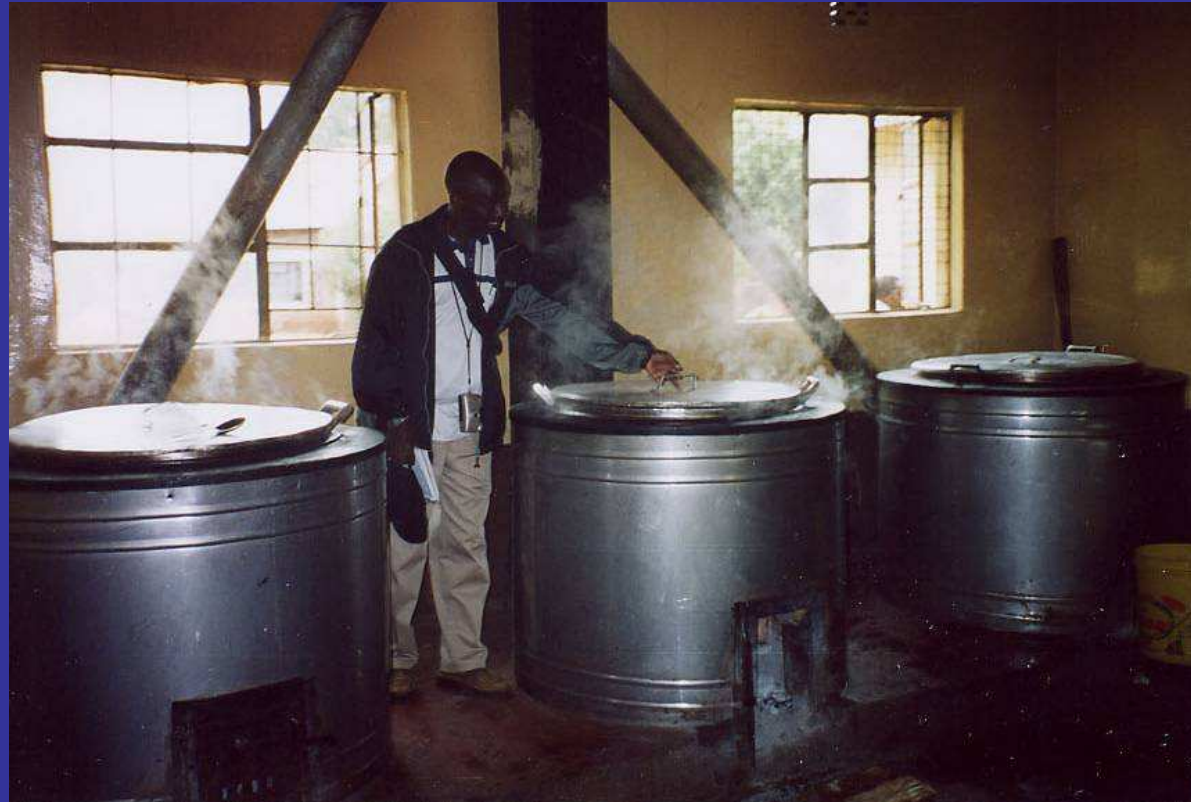
08/30/2003

Central American Griddle Stove

In Central America where tortillas are a major part of the cooking task our griddle stove has been found to save up to 70% or the fuel use



Tea Estates in Africa



In Southern Africa we have institutional sized rocket stoves at tea plantations that are cooking for 40,000 people



A visual comparison between the quantity of wood used (170kg) for the open fire vs. the amount of wood used (13kg) by the 100L Rocket stove. Independently tested by EP Lauderdale Tea Estates (Malawi)

Many other stoves



Many other stoves



Using the shielded fire stove

Many other stoves



Using the rocket – lorena stove

Many other stoves



Many other stoves



Many other stoves



Many other stoves



Many other stoves



Many other stoves

